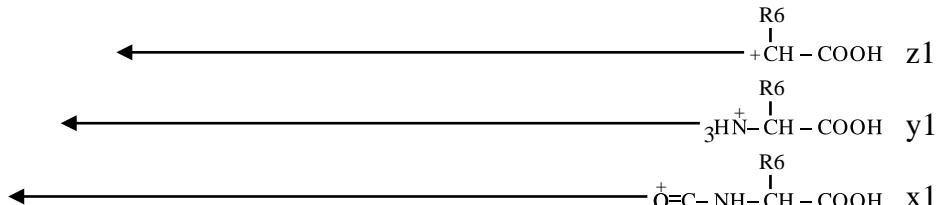


$$m(a1) = m(\text{monomer 1}) - C1O1 + \text{left cap (H1)}$$

$$m(b1) = m(\text{monomer 1}) + \text{left cap (H1)}$$

$$m(c1) = m(\text{monomer 1}) + N1H3 + \text{left cap (H1)}$$



* $m(z1) = m(\text{monomer 6}) - N1H1 + \text{right cap (O1H1)}$ (variant: +H1)

$$m(y1) = m(\text{monomer 6}) + H2 + \text{right cap (O1H1)}$$

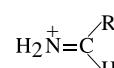
$$m(x1) = m(\text{monomer 6}) + C1O1 + \text{right cap (O1H1)}$$

* Note how a z fragment is identical to a [y -NH3] fragment.

In some cases (high CID energy) the z fragment is often seen as a species of mass $z+1$



canonical monomer



immonium ions